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3 → ___ KCl + ___ O₂ a. 3 b. 5 c. 6 d. 7 e. 8
6. Calculate the mass of hydrogen formed when 27 g of aluminum reacts with excess hydrochloric acid according to the balanced equation below. $2\text{Al} + 6\text{HCl} \rightarrow 2\text{AlCl}_3 + 3\text{H}_2$ a. 1.5 g b. 2.0 g c. 3.0 g d. 6.0 g e. 12 g
7. How many grams of nitric acid, HNO₃, can be prepared from the reaction of 138 ...

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Answer Key and Tests for Discovering Design with Sample ...

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Chemistry - Chapter 3 Test Review Answer all the questions in complete sentences

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(just restate the original question). Show the math work for # 27-30. 1. The law of conservation of mass follows from the concept that _____. 2. If 4 g of element A combine with 10 g of element B, then 12 g of element A combine with ____ g of element B. 3.

Test Bank - Chapter 3 - Middle School Chemistry

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CHAPTER 3 TEST Class Atoms: The Building Blocks of Matter MULTIPLE CHOICE On the line at the left of each statement, write the letter of the choice best completes the statement or answers the question. The behavior of cathode rays in a glass tube containing gas at low pressure led scientists to conclude that the rays were composed of a. energy

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10 Chapter 1 Review Solutions Since mL and cm³ are the same, that means 1.1 L is the same as 1,100 cm³. 20. The answer is 0.1436 m³. We don't have a relationship between L and m³, but we do know that mL and cm³ are the same. Thus, this question requires us to convert from L to mL, then from cm³ to m³.
143,600 mL

9th class chemistry chapter 3 online mcq test with answers ...

Chapter 3 Test Multiple Choice Identify the choice that best completes the statement or answers the question. 1. Most of the volume of an atom is occupied by the a. protons. c. nucleus. b. electron cloud. d. nuclides. 2. When the light from excited atoms of an element is passed through a prism, the distinct colored lines produced are called 3.

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