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Chemistry (4th Edition) Burdge, Julia Publisher McGraw-Hill Publishing Company  
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mass of 3 mol O = 3 (16.00 g) = 48.00 g molar mass of H<sub>2</sub>SO<sub>3</sub> = 82.086 g = 82.09 g  
mass of 2 mol H = 2 (1.008 g) = 2.016 g mass of 1 mol S = 32.07 g = 32.07 g  
mass of 4 mol O = 4 (16.00 g) = 64.00 g molar mass of H<sub>2</sub>SO<sub>4</sub> = 98.086 g = 98.09 g.  
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Steve Zumdahl is the author of market-leading textbooks in introductory chemistry, general chemistry, honors-level chemistry, and high school chemistry. Recently retired from his long-time position as Director of Undergraduate Programs at the University of Illinois, he has received numerous awards for his contributions to chemical education.

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